

# Bio-plastic polymers from renewable sources

## Bio-plásticos de polímeros a partir de fuentes renovables

Polímeros bioplásticos de fontes renováveis

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✉ BAIGTS-ALLENDE DIANA K.(1), PÉREZ-ALVA ALEXA(1), SANDOVAL-HARO MARÍA F.(1),  
SORROZA-MARTÍNEZ ADRIANA P.(1), METRI-OJEDA JORGE C.(1)

(1) DEPARTAMENTO DE INGENIERÍA QUÍMICA, ALIMENTOS Y AMBIENTAL.  
UNIVERSIDAD DE LAS AMÉRICAS, SAN ANDRÉS CHOLULA, PUEBLA. MÉXICO.

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### RESUMEN

Los biopolímeros son útiles en la industria por sus propiedades elásticas y su carácter sustentable en el reemplazo de polímeros no renovables. En este artículo se elaboraron y caracterizaron bioplásticos utilizando quitosano (CH), alginato de sodio (SA) y galactomanano (GAL) a partir de insectos (*Hermetia illucens*), algas pardas (*Macrocystis pyrifera*) y semillas (*Leucaena leucocephala*), respectivamente. La estructura de los biopolímeros se observó por espectroscopía de infrarrojo (FTIR) y se caracterizó en viscosidad a diferentes concentraciones, y los bioplásticos desarrollados se caracterizaron en color y propiedades mecánicas (textura). Los resultados fueron comparados con muestras estándar (comerciales). Los espectros de FTIR confirmaron la presencia de la estructura típica (huella) de los polímeros obtenidos. El SA mostró una viscosidad significativamente mayor en todas las concentraciones comparado con el estándar y los otros polímeros. La fuerza de los bioplásticos fue similar entre CH, SA y GAL para todas las concentraciones; únicamente SA (0,5%) demostró una mayor fuerza que el estándar. Para las mediciones de color, los valores de matiz indicaron colores rojo-amarillento y el croma aumentó proporcionalmente a la concentración de polímero. Las propiedades observadas sugieren que estas fuentes sustentables son una alternativa para la producción de bioplásticos y podría mejorarse su funcionalización por interacciones moleculares para su aplicación en diferentes sectores industriales.

**PALABRAS CLAVE:** galactomanano, quitosano, alginato de sodio.

## ABSTRACT

Biopolymers are useful in the industry due to its elastic properties and sustainability as replacements of non-renewable polymers. In this article, bioplastics were produced and characterized using chitosan (CH), sodium alginate (SA), and galactomannan (GAL) from insects (*Hermetia illucens*), brown algae (*Macrocystis pyrifera*) and seeds (*Leucaena leucocephala*), respectively. The structure of biopolymers was observed by infrared spectroscopy (FTIR) and characterized by viscosity at different concentrations. The developed bioplastics were characterized by color and mechanical properties (texture). The results were compared to standard samples (commercial). The FTIR spectra confirmed the presence of the typical structure (footprint) for the obtained polymers. The SA showed significantly higher viscosity for all concentrations compared to the standard and the other polymers. The bioplastics strength was similar among CH, SA and GAL for all concentrations; only SA (0,5%) demonstrated higher strength than the standard. For color measurements, hue value indicated a red-yellowish color and the chrome increased directly proportional with polymer concentration. The observed properties suggest that these sustainable sources might be an alternative to bioplastic production, which can be extended to functionalization and molecular interactions for broad applications in different industries.

**KEYWORDS:** galactomannan, chitosan, sodium alginate.

## RESUMO

Os biopolímeros são úteis na indústria por suas propriedades elásticas e sua natureza sustentável na substituição de polímeros não renováveis. Neste artigo, os bioplásticos foram preparados e caracterizados utilizando quitosana (CH), alginato de sódio (SA) e galactomanano (GAL) de insetos (*Hermetia illucens*), algas marrons (*Macrocystis pyrifera*) e sementes (*Leucaena leucocephala*), respectivamente. A estrutura dos biopolímeros foi observada por espectroscopia no infravermelho (FTIR) e caracterizada em viscosidade em diferentes concentrações, e os bioplásticos desenvolvidos foram caracterizados em propriedades mecânicas e de cor (textura). Os resultados foram comparados com amostras padrão (comerciais). Os espectros de FTIR confirmaram a presença da estrutura típica (pegada) dos polímeros obtidos. O SA mostrou uma viscosidade significativamente maior em todas as concentrações em comparação com o padrão e os outros polímeros. A força dos bioplásticos foi semelhante entre CH, SA e GAL para todas as concentrações; somente o SA (0,5%) demonstrou maior força que o padrão. Para medições de cores, os valores de matiz indicaram cores vermelho-amareladas e o croma aumentou proporcionalmente à concentração do polímero. As propriedades observadas sugerem que essas fontes sustentáveis são uma alternativa para a produção de bioplásticos, podendo ser melhoradas por interações moleculares para aplicação em diferentes setores industriais.

**PALAVRAS-CHAVE:** galactomanano, quitosana, alginato de sódio.

## INTRODUCTION

Nowadays, concern about the environmental impact of industrial sectors has driven the search for sustainable alternatives for both material production and processing. One of the main problems is plastic waste pollution, due to these materials not being biocompatible and, at the same time, highly resistant to degradation. Bio-based polymers are used as an alternative to replacing non-renewable plastics using available, sustainable, and local sources (Wool and Sun, 2005).

Bio-based polymers can be used as adhesives, foams, films, rigid or flexible plastics, coatings, elastomers, rubber and resins, among others (Wool and Sun, 2005) in several industries such as food, chemical, pharmaceutical, cosmetic, textile and agriculture. Some compounds used for that purpose are starch, cellulose, carrageenan, proteins (gluten, soy, zein), gums, alginate, galactomannan and chitosan, which have been considered as novel ingredients (Verlinden, et al., 2007; Vox, et al., 2013).

Galactomannan is a polysaccharide obtained from legumes endosperm. It is structured by a skeleton of mannoses linked by  $\beta$ - (1,4) bonds and branches formed by galactose units linked by  $\alpha$  (1,6) bond. Galactomannan edible coatings (konjac, *Cassia grandis* and *Gleditsia triacanthos*), as well as guar and locust bean gum, have been used for food and agricultural applications (Mikkonen, et al., 2010; Vox, et al., 2013; Albuquerque, et al., 2017). *Leucaena leucocephala* (guaje), a legume native from Mexico, belongs to family Fabaceae (Ferreira, et al., 2016; Mittal, et al., 2016) and it is rich mainly in protein (31,70%) and carbohydrates (61,31%) (Román-Cortés, et al., 2014; Almeida Filho, et al., 2016). The principal carbohydrate fraction is the soluble fiber galactomannan, which is formed by linear chains of  $\beta$ -(1-4)-D-mannose units substituted by single  $\alpha$ -D-galactose units at O-6 (Ferreira, et al., 2016; Mittal, et al., 2016).

Another highly available biopolymer is chitosan, which is obtained from chitin (N-acetyl glucosamine with  $\beta$ -(1,4) bonds) deacetylation. It is present in crustaceans, insects and other species such as yeast and fungi. Chitosan has been applied in multiple areas (food, cosmetics, textile and pharmaceutical) to produce ecological materials (Caligiani, et al., 2018; Chávez, et al., 2019) such as edible coatings, antimicrobial films and large-scale 3D products (Fernández and Ingber, 2014; Revathi and Thambidurai, 2017; Souza, et al., 2017).

Recently, increasing interest in obtaining compounds from insects such as protein, oil, and polysaccharides has taken place because they are a sustainable and low-cost source for raw material production (Caligiani, et al., 2018). From this point of view, the inclusion of emerging materials such as microalgae, duckweeds (aquatic plants) and algae is being encouraged for the production of added-value compounds. An example is macroalgae, which is rich in alginic acid, also known as alginate. Its salts, especially sodium alginate, are useful for industrial applications such as food packaging material, paper, coating and also for pharmaceutical and medical purposes. In the food industry, it is used as an edible coating and bio-active compounds carrier (De'Nobili, et al., 2016; Ferreira, et al., 2016). Sodium

alginate is a polyelectrolyte constituted by monomeric units of  $\alpha$ -D-mannuronate and  $\beta$ -l-glucuronate linked by 1-4 bonds. The brown algae (“giant kelp” *sargassum*) has recently become an environmental concern because of its accumulation in the sea and the harmful effect it causes to a wide array of animal species and coral reefs (Popa, et al., 2011).

This study is aimed to utilize three different sustainable sources (insects, legume seed and brown algae) to obtain bio-based polymers as a potential alternative for bioplastics production.

## MATERIALS AND METHODS

### Materials

PROENTO Company kindly provided black soldier fly larvae (*Hermetia illucens*) (Queretaro, Mexico), brown algae (*Macrocystis pyrifera*) was purchased from ALGAS PACIFIC (Baja California, Mexico), and guaje seeds (*Leucaena leucocephala*) were obtained from a local market in Oaxaca, Mexico. HCl, NaOH, acetic acid, CaCl<sub>2</sub>, Na<sub>2</sub>CO<sub>3</sub>, ethanol, and glycerol were analytical grade acquired by Sigma-Aldrich Co. Commercial chitosan, sodium alginate and xanthan gum were purchased from Sigma-Aldrich Co and FMC Corporation, respectively, and used as standard polymers.

### Chitosan production

Black soldier fly larvae chitin was demineralized using HCl 1N in a ratio of 1:15 (w/v) for 1 hour at 40 °C and continuously stirred, the mixture was washed with distilled water until its neutralization. Next, proteins were removed with NaOH (1M) solution in a ratio of 1:15 (w/v) at the same conditions. Chitin was discolored using an ethanol/acetone solution (1:1) at a ratio of 1:15 (w/v) under constant stirring for 1 hour. The treated chitin was dried by convection at 40 °C. For chitosan production, chitin was deacetylated by boiling chitin solutions (1:15 w/v) in NaOH (50%) for 1 hour. Lastly, chitosan was washed using distilled water until neutrality and dried overnight at 30 °C (Song, et al., 2018).

### Sodium alginate production

Brown algae powder (*Macrocystis pyrifera*) was soaked overnight (20 h) in 600 mL of CaCl<sub>2</sub> (1%) at room temperature. Then the algae were washed with distilled water (3 times), afterwards, the pellet was resuspended in 250 mL of HCl 5% for one hour, filtered and then re-washed with distilled water. The obtained pellet was incubated overnight in 600 mL of Na<sub>2</sub>CO<sub>3</sub> (3%). The mixture was centrifuged at 6000 rpm for 20 min at room temperature. Sodium alginate was precipitated from the supernatant by adding ethanol (1:1 v/v). Finally, sodium alginate was washed with an excess of ethanol and water and freeze-dried (Chee, et al., 2010).

## Galactomannan production

Dried guaje seeds (*Leucaena leucocephala*) were soaked in water (1:10) and stirred at 60 °C for 1 hour. Afterward, the mixture was filtered, and the extract was centrifuged at 6000 rpm at 25 °C for 5 minutes. The galactomannan was precipitated by adding ethanol 96% to the supernatant (1:3 v/v) under slow and constant stirring, filtrated, and freeze-dried (Mittal, et al., 2016).

## Fourier transform infrared spectroscopy (FTIR)

The structure of the biopolymers chitosan (CH), galactomannan (GAL) and sodium alginate (SA) was determined by FTIR and compared to commercial samples. Dried-samples (powder) footprint was obtained in a range of 400-4000 cm<sup>-1</sup> using an Agilent Cary 630 FTIR infrared spectrometer (Agilent technologies).

## Viscosity

Rheological properties of biopolymer aqueous dispersions at three concentrations (0,5%, 1% and 1,5%) were measured using a Brookfield Viscometer DV-II equipped with an SC-27 needle at 20 °C at 100 rpm. Galactomannan and sodium alginate dispersions were heated at 60 °C x 1 h and 40 °C x 10 min (respectively) and cooled down to room temperature prior measurements. In the case of chitosan, the used solvent was acetic acid (0,1 M) and the mixture was kept under stirring overnight. Commercial chitosan, xanthan gum and sodium alginate were used as control samples.

## Bioplastics preparation

Biopolymer solutions (0,5%, 1% and 1,5%) with 1% of glycerol were stirred until its homogenization, then 15 mL of the mixture was poured in a silicone cast and left to dry at room temperature for approximately 48 hours.

## Bioplastic characterization

The color of bioplastics was determined using a Konica Minolta CR-400 Colorimeter. The results obtained were expressed in CIELab coordinate system. Chroma (C\*) and hue angle (H\*) were calculated using Equation 1 and Equation 2, respectively.

$$C^* = \sqrt{a^{*2} + b^{*2}} \quad (\text{Eq. 1})$$

$$H^* = \tan^{-1} \left( \frac{b^*}{a^*} \right) \quad (\text{Eq. 2})$$

The mechanical properties of bioplastics were measured using a texture analyzer (Shimadzu EZ-SX). Uniaxial compression test was performed to study the fracture properties for a better understanding of the energy in the matrix and determine

the type of network formed by the studied polymers. The bioplastics were cut into cylindrical samples (12 mm of diameter and 0,5 mm of thickness), the final distance was set at 3 mm of deformation using 1 mm/s of speed at room temperature. Data was analyzed by Trapezium software (TRAPEZIUMX).

## Statistical analysis

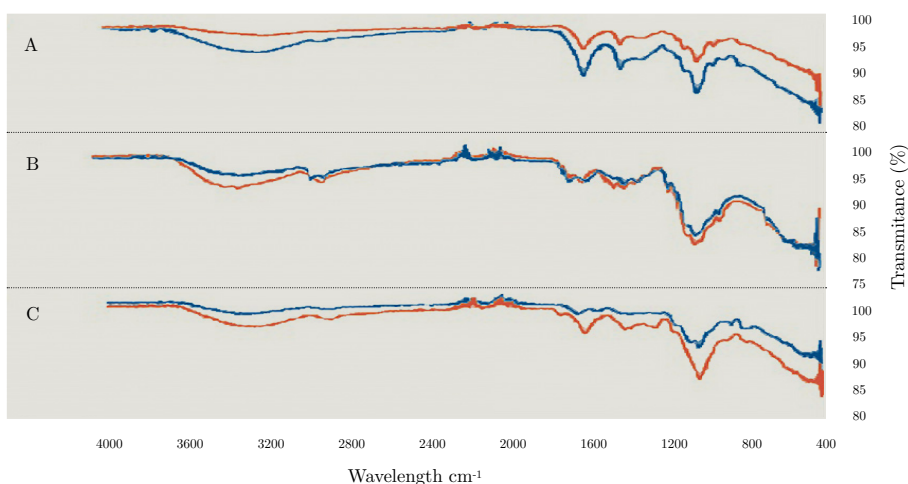
All experiments were done in triplicate ( $n = 3$ ) and an analysis of variance (ANOVA) of the data was performed using the Minitab software version 18. Significant differences were determined by using Duncan's multiple comparison procedure at  $p < 0,05$ .

## RESULTS

### FTIR footprint

Extracted biopolymers showed a similar footprint than standard samples for all cases. However, in some regions, higher intensity peaks were observed (Figure 1). In the case of SA, standard samples showed a lower presence of hydroxyl, manuronic and uronic acid groups ( $2800-3600\text{ cm}^{-1}$  and  $1150-800\text{ cm}^{-1}$ ) compared to the extracted biopolymers (Figure 1A). For chitosan (Figure 1B), higher peaks for the standard samples were observed uniquely at  $2800-3600\text{ cm}^{-1}$  region (C-H, N-H, and O-H bonds). In Figure 1C, it can be observed that xanthan gum (standard) showed an increased intensity compared to GAL, mainly at  $900-1150\text{ cm}^{-1}$   $800-870\text{ cm}^{-1}$  (C-O and C-O-H groups) in the carbohydrate structure.

These differences demonstrated some structural changes in the biopolymers; it is well known that the molecule structure is determined by several factors such as the extraction parameters, the nature of the raw material, standardization (commercial material), among others. Commercial chitosan and xanthan gum (CH standard and GAL standard, respectively) were obtained from crustaceous and bacteria, respectively, unlike the biopolymers from this study, which were obtained from insects and legume seed. In the case of SA, even though both standard and extracted are from brown algae, the compositional characteristics and extraction method could differ from each other affecting its structure.



**Figure 1.** FTIR spectra for SA (A), CH (B), and GAL (C) in blue lines and their respective standard polymers in orange lines.

## Viscosity

Flow behavior of biopolymers solutions is shown in Table 1. For all cases, the solutions behaved as shear-thinning fluids (viscosity decreases with increasing shear stress), and the viscosity increased with the biopolymer concentration. At all concentrations, SA showed the highest viscosity values ( $\sim 101$  cP, 487 cP, and 1,212 cP for each concentration) compared to the other samples, including the standard. In contrast, CH and GAL presented lower viscosity ( $p < 0,05$ ) compared to their respective standards at all concentrations.

**Table 1.** Extracted SA, CH, GAL and control polymers viscosity (cP) at 0,5, 1 and 1,5% concentrations.

Polymer	Viscosity (cP)		
	0,50%	1%	1,50%
SA	101,25 ± 1,767	487,5 ± 7,071	1212,5 ± 10,606
SA standard	21,67 ± 1,443	91,67 ± 1,443	244,17 ± 5,204
CH	4,83 ± 0,288	12,33 ± 0,288	15,16 ± 0,288
CH standard	32,5 ± 3,530	102,5 ± 0,001	467,5 ± 0,001
GAL	12,5 ± 0,000	120 ± 3,535	393,75 ± 1,767
GAL standard	392,5 ± 17,677	705 ± 21,213	922,5 ± 45,961

## Bioplastics characteristics: color and texture analysis

Standard bioplastics colors were statistically different ( $p < 0,05$ ) compared to the obtained from the extracted polymers, in general colorless and more translucent

films were observed (Table 2). SA, CH, and GAL bioplastics showed a hue value range from 80 to 100°, indicating a red-yellowish color. Chroma (color saturation) increased proportionally with polymer concentration, the higher values were for SA ( $15,74 \pm 0,48 - 25,06 \pm 0,21$ ), followed by GAL ( $12,65 \pm 0,14 - 23,00 \pm 0,02$ ) and CH ( $8,63 \pm 0,37 - 12,01 \pm 0,74$ ).

**Table 2.** Color (L, a\*, b\* chroma and hue) of bioplastics obtained by different polymers at 0,5, 1 and 1,5%.

Polymer (%)	L	a*	b*	Chroma	Hue	
CH	0,5	$91,98 \pm 0,304$	$-0,65 \pm 0,068$	$8,60 \pm 0,160$	$8,63 \pm 0,374$	$94,47 \pm 0,374$
	1	$88,59 \pm 1,449$	$-0,48 \pm 0,026$	$10,25 \pm 0,257$	$10,26 \pm 0,190$	$92,68 \pm 0,190$
	1,5	$89,25 \pm 0,156$	$-0,43 \pm 0,156$	$12,01 \pm 0,051$	$12,01 \pm 0,742$	$92,06 \pm 0,742$
CH standard	0,5	$92,09 \pm 0,490$	$-1,11 \pm 0,045$	$8,40 \pm 0,005$	$8,47 \pm 0,303$	$97,54 \pm 0,303$
	1	$93,03 \pm 0,010$	$-1,31 \pm 0,020$	$8,57 \pm 0,040$	$8,67 \pm 0,111$	$98,71 \pm 0,111$
	1,5	$93,56 \pm 0,037$	$-1,24 \pm 0,046$	$8,58 \pm 0,035$	$8,67 \pm 0,329$	$98,23 \pm 0,329$
SA	0,5	$86,3 \pm 1,019$	$-0,17 \pm 0,051$	$15,74 \pm 0,489$	$15,74 \pm 0,202$	$90,63 \pm 0,202$
	1	$84,82 \pm 0,650$	$0,39 \pm 0,145$	$20,26 \pm 0,618$	$20,26 \pm 0,388$	$88,89 \pm 0,388$
	1,5	$79,91 \pm 0,290$	$1,03 \pm 0,098$	$25,04 \pm 0,212$	$25,06 \pm 0,204$	$87,63 \pm 0,204$
SA standard	0,5	$94,08 \pm 0,060$	$-0,98 \pm 0,010$	$5,69 \pm 0,122$	$5,77 \pm 0,145$	$99,77 \pm 0,145$
	1	$93,25 \pm 0,540$	$-1,02 \pm 0,045$	$6,79 \pm 0,331$	$6,86 \pm 0,046$	$98,59 \pm 0,046$
	1,5	$92,33 \pm 0,213$	$1,08 \pm 0,078$	$8,44 \pm 0,786$	$8,51 \pm 0,237$	$97,29 \pm 0,237$
GAL	0,5	$90,17 \pm 0,070$	$0,23 \pm 0,014$	$12,65 \pm 0,148$	$12,65 \pm 0,148$	$88,95 \pm 0,076$
	1	$87,72 \pm 0,100$	$1,32 \pm 0,176$	$19,44 \pm 0,289$	$19,49 \pm 0,277$	$86,09 \pm 0,576$
	1,5	$84,99 \pm 0,014$	$2,32 \pm 0,007$	$22,88 \pm 0,021$	$23,00 \pm 0,021$	$84,19 \pm 0,012$
GAL standard	0,5	$93,6 \pm 0,304$	$-0,53 \pm 0,014$	$4,71 \pm 0,091$	$4,74 \pm 0,092$	$96,41 \pm 0,045$
	1	$93,59 \pm 0,806$	$-1,21 \pm 0,007$	$6,78 \pm 0,021$	$6,89 \pm 0,019$	$100,15 \pm 0,088$
	1,5	$94,27 \pm 0,098$	$-1,03 \pm 0,014$	$6,27 \pm 0,728$	$6,35 \pm 0,716$	$99,38 \pm 1,196$

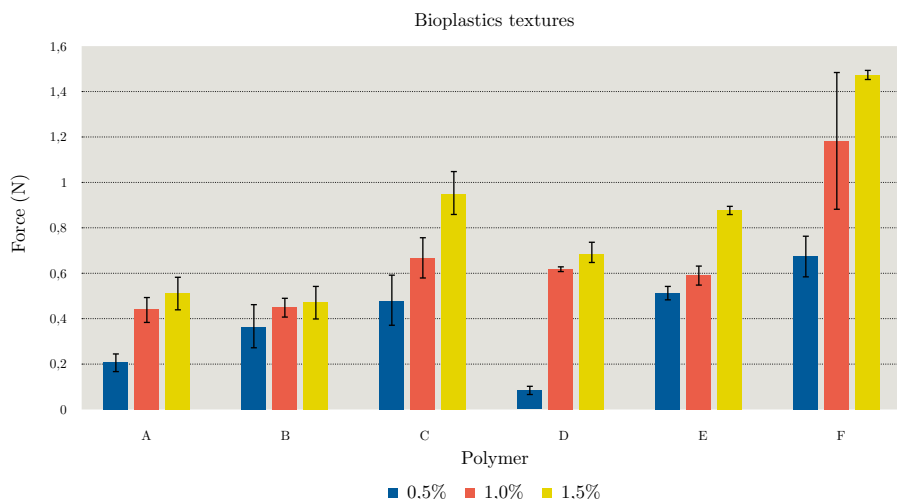
## Texture

The highest strength observed was for GAL bioplastics and its standard for all concentrations measured ( $0,955 \pm 0,091$  and  $1,478 \pm 0,026$  N, respectively). However, standard bioplastics were stronger. Similar results were found for CH at 0,5 and 1% ( $0,369 - 0,451$  N) in comparison to standard bioplastics. At 1,5% of CH ( $0,475 \pm 0,021$  N) the firmness was significantly lower ( $p < 0,05$ ) than standard ( $0,881 \pm 0,069$  N).

SA films showed a significantly higher mechanical resistance ( $p < 0,05$ ) compared to its standard ( $0,207 \pm 0,038$  and  $0,085 \pm 0,020$  N, respectively), nevertheless,



when the concentration of the polymers increased, SA bioplastics presented lower resistance at 1 and 1,5 % ( $p < 0,05$ ).



**Figure 2.** Mechanical properties of bioplastics from sodium Alginate (A), chitosan (B), galactomannan (C) and their controls (D, E and F, respectively).

## DISCUSSION

The functionality of biopolymer molecules is closely related to its structure; thus, the presence and characteristics of the structural elements will determine its usage. In general, the structures observed by FTIR were in agreement with the flow behavior (viscosity).

SA samples showed a more preserved structure (higher absorbance peaks) denoted by its higher viscosity in comparison to standard samples. Probably a higher proportion of mannuronic and uronic acids ( $800-1150\text{ cm}^{-1}$ ) allowed better structural building blocks formation and water interactions in the case of viscosity (Zubia, et al., 2008; Gomez, et al., 2009). In GAL polymer, although similar footprint was obtained (compared to its standard), the chemical differences observed were reflected on the viscosity of the dispersions and bioplastics texture. However, the viscosity of legume seed GAL was considerably higher ( $\sim 120\text{ cP}$ ) than reported values for the same raw material ( $15\text{ cP}$ ) at equal concentrations (Mittal, et al., 2016). Non-structural changes were observed for CH biopolymers; nevertheless, the viscosity varied greatly. A critical parameter in these types of compounds is the degree of deacetylation, which in large part determine its molecular weight. The deacetylation degree of CH from insects in this work was not determined. However, commercial chitosan (from shrimp shells) was between 75-85% (medium molecular weight); these features may be the leading cause of distinct viscosity and texture. Bioplastics color depends on the type of source and the treatment applied. In this study, the extracted polymers were not previously purified. Therefore the presence of pigments

(chlorophylls, carotenoids and polyphenols) could decrease the clearness of the obtained bioplastics.

The strength of bioplastics produced using CH (~30,74 – 39,27 MPa), SA (~17,25 – 44,05 MPa) and GAL (~40,28 – 79,58 MPa) in this study was similar than the reported values from different authors (25,3 – 39,3 MPa, 22,5 – 35,0 MPa and 25 – 40 MPa), respectively (Galus and Lenart, 2013; Moura, et al., 2015; Jussen, et al., 2019).

## CONCLUSIONS

The desired functional properties of bio-based polymers suggest that the non-conventional sustainable sources from insects (*Hermetia illucens*), legume seeds (*Leucaena leucocephala*) and brown algae (*Macrocystis pyrifera*) could be an alternative for bioplastic production, which can be extended to functionalization and molecular interactions for broad applications in different industrial sectors.

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